1.4-Additions of the Highly Functionalized Copper Reagents RCu(CN)ZnI · 2 BF3 to Trisubstituted Enones.

A New BF3 Promoted Cyclization Reaction

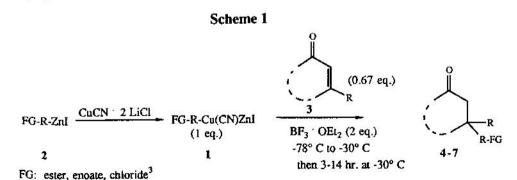
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Summary: The new copper reagents  $RCu(CN)Zn\ I \cdot 2\ BF_3\ 1$  containing various functional groups like esters, vanides, chlorides or enoates react with  $\beta$ ,  $\beta$ - disubstituted enones to afford the Michael adducts 4-7 in high yields \$1.98%). Cyano-substituted 1,4- adducts undergo a new cyclization reaction leading to the very stable bicyclic bifluoroboron enolates 11a-11c which could be converted into the bicyclic diketones 12 under mild basic conditions. The X-ray structure of 11b is reported.

Recently, we reported  $^1$  that the highly functionalized copper, zinc organometallics FG-R-Cu(CN)ZnI  $^1$  obtained from the corresponding alkylzinc iodides  $^2$  react with various electrophiles like acyl chlorides, allylic halides, aldehydes or chones to afford polyfunctional molecules in high yields. However a  $\beta$ ,  $\beta$ '- disubstituted enone like 3-methyl-2-cyclohexen-1-one gave the desired  $^1$ ,4- adduct in only a moderate yield  $^1$ a.

We wish now to report that the presence of BF<sub>3</sub> · OEt<sub>2</sub><sup>2</sup> allows us to perform Michael additions of the organo-copper reagents 1 to highly substituted enones 3 in excellent yields (81-98%). See Scheme 1 and Table 1.



In a typical procedure, 2.5 ml (20 mmol) of BF<sub>3</sub> · OEt<sub>2</sub> was added at -78° C to a THF solution of 10 mmol of the copper reagent 1, followed by 6.67 mmol (0.67 eq.) of the enone 3. The reaction mixture was allowed to reach

<sup>3</sup>0° C and was stirred 3 to 14 hours at this temperature. An additional 2 hours of stirring at 0° C was necessary in some

cases. (See Table I). After the usual work-up, the resulting oil was purified by flash chromatography. Since  $_{no~1.4}$  addition reaction occurs without the presence of BF<sub>3</sub> · OEt<sub>2</sub>, a selective synthesis of highly functionalized enones of type 8 and of the  $\beta$ ,  $\beta$ '- disubstituted ketones 7a-c was possible respectively by the reaction of 3-iodo-2-cyclohexen. 1-one<sup>4</sup> 9 with R<sup>1</sup>Cu(CN)Zn I (THF, 3 hrs., -30° C; 78-89%) and by the reaction of enones 8 with R<sup>2</sup>Cu(CN)Zn I (2 BF<sub>3</sub> (see Scheme 2 and Table I).

Noteworthy in all these BF3-promoted reactions is the tolerance of various oxygen containing functional groups. 1,4- Adducts containing a cyano group show a different behavior. After the initial Michael addition leading to 10, a ring closure reaction occurs and affords the bicyclic diffuoroboron enolates 11 which could be isolated in good yields (79-82%) and converted under mild basic conditions ( (i) MeONa/MeOH, 0° C, 30 min. (ii) SiO<sub>2</sub> (flash chromatography)) to the bicyclic ketones 12 in 71-78% yield (see Scheme 3).

## Scheme 3

The boron enolates 11 are remarkably stable compounds which could be purified by flash chromatography (solvent: AcOEt / Hexane). An X-ray analysis<sup>5</sup> of 11b ( $R = -(CH_2)_3CN$ ) disclose a very symmetrical structure with a C-N double bond (1.31 Å) and with a C10-C9 bond and a C9-O bond of 1.36 Å and 1.33 Å length respectively which is typical for an enolate<sup>6</sup>. Interestingly the tetrahedral boron is not in the plane of the organic chelate but is slightly bent toward the same side as the substituent at C5 (see Fig. 1).

Table I. Products 4-7 Obtained by the Reaction of Enones with Organo-Copper Reagents 1

Yield(%)a Products of 4-7 Enone RCu(CN)Zn I 1 4a:  $R = -(CH_2)_4 - CI$ 88 Cl-(CH2)4-Cu(CN)ZnI 98b 4c:  $R = -(CH_2)_3OAc$ 87 AcO-(CH2)3-Cu(CN)ZnI 5a:  $R = -(CH_2)_4 - C1$ 94b,c Cl-(CH2)4-Cu(CN)ZnI **5b:**  $R = -(CH_2)_3CO_2Et$ 94b,c EtO2C-(CH2)3-Cu(CN)ZnI 6a:  $R = -(CH_2)_2 - CO_2E_1$ 88 EtO2C-(CH2)3-Cu(CN)ZnI 88b **6c:**  $R = -(CH_2)_3 - CH(CH_3) - OPiv$ 88b PivO-CH(CH3)-(CH2)3-Cu(CN)ZnI 81d Cl-(CH<sub>2</sub>)<sub>4</sub>Cu(CN)ZnI 86b,c PivO-CH(CH3)-(CH2)3-Cu(CN)ZnI 86 AcO-(CH2)3-Cu(CN)ZnI

All indicated yields are isolated yields. Satisfactory spectral data (IR, <sup>1</sup>H and <sup>13</sup>C-NMR, mass spectra), high resolution mass spectra or elemental analysis were obtained for all compounds. The reaction was stirred I4 hrs. at -30° C then 2 hr. at 0° C. A 1:1 mixture of diastereoisomers was obtained. The reaction was stirred 22 hrs. at -30° C, then 6 hrs. at 0° C. d

m.p. 177-179°C

<sup>1</sup>H-NMR: δ(H-N) 6.85 ppm

<sup>13</sup>C-NMR: δ(CI) 172.4 ppm

δ (C10) 111.5 ppm

δ (C9) 175.90 ppm

IR (CDCl<sub>3</sub>): v (NH) 3147 cm<sup>-1</sup> v (C=C or C=N) 1614 cm<sup>-1</sup>

Elemental analysis:

1.

2.

5.

Calcd: C: 60.03; H: 6.81; N: 10.00

Found: C: 60.02; H: 6.84; N: 9.99

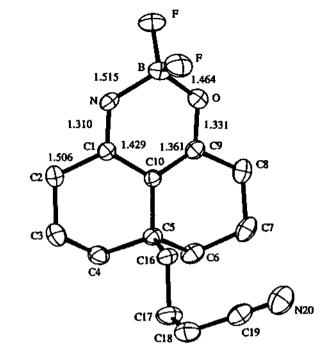


Fig. 1. ORTEP representation of the structure of 11b determined by X-ray analysis and significant analytical data of 11b.

Further synthetic applications of this new BF3-mediated cyclization are currently underway in our laboratory.

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References

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- Crystallographic data for compound 11b: crystal system: orthorhombic space group: Pbca [a = 7.069(2) Å, b= 16.123(5) Å, C=24.497(8) Å ]  $D_{calc.} = 1.333 \text{ g/cm}^3$ ;  $\lambda$  (Mo K $\alpha$ ) = 0.71069 Å; V = 2792 (1) Å<sup>3</sup>; Z = 8; crystal size (mm): 0.23 x0.208 x 0.426 no. of data,  $I > 3\sigma$  (I): 951; no. of refined parameters: 184; R = 0.040;  $R_W = 0.035$ .
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