# Construction of Fused Bicyclo-[5.3.0]decane and -[5.4.0]undecane Ring Skeletons via Sequential Additions of Nucleophiles to Tricarbonyltropyliumchromium Tetrafluroborate 

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The addition of a variety of the highly functionalized zinc-copper reagents $\mathrm{RCu}(\mathrm{CN}) \mathrm{Znl}$ to the title cation gives ( $\eta^{6}$-cyclohepta-1,3,5-triene) $\mathrm{Cr}(\mathrm{CO})_{3}$ complexes with a functionalized side chain at the $\mathrm{C}-7$ position of the ring: intramolecular cyclisation of ester-substituted adducts using lithium diisopropylamide generates fused bicyclo-[5.3.0]decane and -[5.4.0]undecane derivatives.

Seven-membered carbocycles are often used in the construction of complicated natural products. ${ }^{1}$ Specially, the perhydroazulene or bicyclo[5.3.0]decane ring system is the most commonly encountered of the cycloheptane derived structure in natural occurring biologically-active compounds. ${ }^{2}$ Recently, tropone has been shown to be a versatile building block for the synthesis of a variety of sesquiterpenes with the bicyclo[5.3.0]decane ring skeleton. ${ }^{3}$ Normally, several steps starting from the 1,8 -addition of a protected functionalized three-carbon chain to tropone are needed to construct the bicyclo[5.3.0]decane ring skeleton. However, bicyclo[5.3.0]decane and -[5.4.0]undecane derivatives may be available easily via sequential additions of nucleophiles to [(tropylium $) \mathrm{Cr}(\mathrm{CO})_{3}$ ] cation 1. This method allows the formation of 5,7 - and 6,7 -cis-fused bicyclic systems which cannot be generated by sequential additions of nucleophiles to

a; $\mathrm{R}=\left[\mathrm{CH}_{2}\right]_{2} \mathrm{CO}_{2} \mathrm{Et}, 96 \%$
b; $\mathrm{R}=\left[\mathrm{CH}_{2}\right]_{3} \mathrm{CO}_{2} \mathrm{Et}, 96 \%$
c; $\mathrm{R}=\left[\mathrm{CH}_{2}\right]_{4} \mathrm{CO}_{2} \mathrm{Et}, 88 \%$
d; $\mathrm{R}=\left[\mathrm{CH}_{2}\right]_{2} \mathrm{CN}, 69 \%$
e; $\mathrm{R}=\left[\mathrm{CH}_{2}\right]_{3} \mathrm{CN}, 69 \%$
$\mathbf{f} ; \mathrm{R}=\left[\mathrm{CH}_{2}\right]_{4} \mathrm{CN}, 96 \%$
g; $\mathrm{R}=\mathrm{CH}_{2} \mathrm{Ph}, 89 \%$
h; $\mathrm{R}=\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}, 84 \%$
Scheme 1


Fig. 1
( $\eta^{5}$-pentadienyl) $\mathrm{Fe}(\mathrm{CO})_{3} .^{4}$ Prior to our investigation, several classes of nucleophiles such as hydrides, methoxides and hydrogen sulfides had been shown to add to cation 1 to produce 7 -exo-substituted ( $\eta^{6}$-cyclohepta-1,3,5-triene)$\mathrm{Cr}(\mathrm{CO})_{3}$ complexes. ${ }^{5}$ However, the addition of an excess of sodium diethylmalonate or lithium cyclopentadienide to 1 provided the ring contraction product $\left[\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{Cr}(\mathrm{CO})_{3}\right]^{6}$ and the reaction of cyanide with 1 led by reductive coupling to a mixture of bi(cycloheptadienyl) $\mathrm{Cr}_{2}(\mathrm{CO})_{6}$ and bi(cycloheptadienyl) $\mathrm{Cr}(\mathrm{CO})_{3} .{ }^{7}$ Moreover, phenyl lithium failed to react with 1. Surprisingly, reports on the addition of organocopper derivatives to cation 1 are lacking. We now report that the highly functionalized zinc-copper reagents $[\mathrm{RCu}(\mathrm{CN}) \mathrm{ZnI}]^{4,8}$ add efficiently to 1 to afford ( $\eta^{6}$-cyclohepta-1,3,5-triene)$\mathrm{Cr}(\mathrm{CO})_{3}$ complexes 2a-h with a functionalized side chain at the $\mathrm{C}-7$ position of the ring. The reactions are performed in the expectation that further manipulation of the resulting complexes will then be possible after the initial additions.

Cation 1 was prepared in two steps from cyclohepta-1,3,5triene and $\mathrm{Cr}(\mathrm{CO})_{3}(\mathrm{MeCN})_{3}\left(85 \%\right.$ overall yield). ${ }^{9}$ Finally, our synthesis of C -7-substituted ( $\eta^{6}$-cyclohepta-1,3,5-triene)$\mathrm{Cr}(\mathrm{CO})_{3}$ complexes 2 involved the slow addition of a THF solution of the highly functionalized zinc-copper reagents $\mathrm{RCu}(\mathrm{CN}) \mathrm{ZnI}$ (1.2 equiv.) to a stirred suspension of cation 1 in THF at $0^{\circ} \mathrm{C}$ under nitrogen. The addition was carried out for 5 h at $23^{\circ} \mathrm{C}$ followed by workup with saturated aqueous ammonium chloride and diethyl ether extraction. After purification by flash column chromatography on silica gel, complexes 2a-h were obtained as the major products and in good yields ( $69-96 \%$, Scheme 1). The trans relationship between the nucleophile and the metal moiety of complexes 2 was assigned based upon comparison of their ${ }^{1} \mathrm{H}$ NMR spectra [ $\delta 5.95\left(2 \mathrm{H}\right.$, br.s)] with the spectra of known 7-exo- $\left(\eta^{6}-\right.$ cyclohepta-1,3,5-triene) $\mathrm{Cr}(\mathrm{CO})_{3}$ complexes. ${ }^{10}$ Further manipulation of the resulting complexes 2 was demonstrated as follows. Treatment of $\mathbf{2 b}$ with lithium diisopropylamide (LDA, 1.2 equiv.) in THF and hexamethylphosphoramide (HMPA, THF-HMPA $=3: 1$ ) at $-78^{\circ} \mathrm{C}$ for 1 h followed by quenching the reaction mixture with TFA ( 10 equiv.) produced a mixture of perhydroazulene derivatives $\mathbf{3 a}$ and $\mathbf{3 b}$ in a 1:1 ratio in 77\% total yield. The formation of bicyclo[5.3.0]decane derivatives 3a and 3b agrees closely with the mechanism proposed for the intermolecular addition of nucleophiles to ( $\eta^{6}$-arene) $\mathrm{Cr}(\mathrm{CO})_{3} .{ }^{11}$ Deprotonation of $\mathbf{2 b}$ using LDA at $-78^{\circ} \mathrm{C}$ in THF-HMPA would result in the formation of the kinetic ester enolate. ${ }^{4,12}$ Anti addition of the ester enolate to the terminal position of the triene ligands gave ( $\eta^{5}$-cycloheptadienyl) $\mathrm{Cr}(\mathrm{CO})_{3}$ anion complex 4 ( $n=1$, Fig. 1). Electrophilic quenching of 4 with TFA afforded bicyclo[5.3.0]decane derivatives 5a and 5b. However, diene migration occurred and produced thermally more stable isomers $\mathbf{3 a}$ and $\mathbf{3 b}$ under the reaction conditions. Treatment of $2 \mathrm{c} \dagger$ with LDA ( 1.2 equiv.) in THF-HMPA ( $3: 1$ ) under at atmosphere of CO at $-78^{\circ} \mathrm{C}$ for 1 h followed by quenching the reaction mixture with iodomethane ( 5 equiv.; $\mathrm{CO} ; 25^{\circ} \mathrm{C} ; 2 \mathrm{~h}$ ), and then iodine ( 4 equiv.; $25^{\circ} \mathrm{C} ; 4 \mathrm{~h}$ ) generated the bicyclo[5.4.0]undecane derivative $\mathbf{6} \dagger$ as the only diastereo-
isomer isolated in $63 \%$ yield. Reaction of $4(n=2)$ with iodomethane under an atmosphere of CO would give the bicyclo[5.4.0]undecane derivative 713 which isomerized to provide the highly conjugate dienone 6. The cis stereochemistry at the ring juncture of 6 is fixed by anti addition of the ester enolate to the terminal position of the triene ligand (to give $\mathbf{4}, n=2$ ) and the ester group assigned as exo was resulted from the formation of the kinetic ester enolate under the kinectically controlled reaction conditions. ${ }^{4}{ }^{1} \mathrm{H}$ NMR studies provided the initial evidence for support of the structural assignments. The ${ }^{1} \mathrm{H}$ NMR spectrum of compound 6 exhibited a doublet of doublets, centred at $\delta 3.2, J=12.0,2.8$ Hz , assigned to $\mathrm{H}_{\mathrm{a}}$. The coupling constant of $\mathrm{H}_{\mathrm{a}}-\mathrm{H}_{\mathrm{b}}\left(J_{\mathrm{ab}}\right)$ of 12.0 Hz agrees closely with the 12.7 Hz coupling constant observed for the cis fused bicyclo[5.4.0]undecanone and the coupling constant of $\mathrm{H}_{\mathrm{a}}-\mathrm{H}_{\mathrm{c}}\left(J_{\mathrm{ac}}\right)$ of 2.8 Hz agrees with the 3.2 Hz coupling constant observed for the similarly disposed trans hydrogens. ${ }^{14}$

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## Footnote

$\dagger$ Spectroscopic data for: 2c: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 5.95(2 \mathrm{H}, \mathrm{brs}), 4.77$ $(2 \mathrm{H}, \mathrm{brs}), 4.10(2 \mathrm{H}, \mathrm{q}, J 7.2 \mathrm{~Hz}), 3.70(2 \mathrm{H}, \mathrm{t}, J 8.6 \mathrm{~Hz}), 2.90(1 \mathrm{H}, \mathrm{m})$, $2.17(2 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}), 1.39(2 \mathrm{H}, \mathrm{m}), 1.23(3 \mathrm{H}, \mathrm{t}, J 7.2 \mathrm{~Hz}), 1.06(2 \mathrm{H}$, $\mathrm{m})$ and $0.25(2 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 231.9,173.4,98.9,97.8$, $67.7,60.2,38.6,35.9,34.0,24.3,23.2$ and 14.2; IR $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$, solution) $\mathrm{v} / \mathrm{cm}^{-1} 3067 \mathrm{w}, 2992 \mathrm{w}, 1977 \mathrm{~s}, 1910 \mathrm{~s}, 1879 \mathrm{~s}, 1728 \mathrm{w}, 1620,1424 \mathrm{~m}$ and 1252m; mass (EI) m/z 357 ( $\mathrm{M}^{+}, 27 \%$ ), 328 (8), 300 (5), 272 (73), 254 (11), 194 (46), 180 (100); high resolution, calc. for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{O}_{5} \mathrm{Cr}$ 356.0710; found $m / z 356.0712$. For 6: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 6.84(1 \mathrm{H}, \mathrm{d}$, $J 7.8 \mathrm{~Hz}), 6.14(1 \mathrm{H}, \mathrm{dd}, J 10.7,8.3 \mathrm{~Hz}), 5.91(1 \mathrm{H}, \mathrm{m}), 4.14(2 \mathrm{H}, J 6.8$ $\mathrm{Hz}), 3.26(1 \mathrm{H}, \mathrm{dd}, J 12.0,2.8 \mathrm{~Hz}), 2.71-2.61(2 \mathrm{H}, \mathrm{m}), 2.44(1 \mathrm{H}, \mathrm{m})$, $2.34(3 \mathrm{H}, \mathrm{s}), 2.12(1 \mathrm{H}, \mathrm{m})$ and $1.82-1.23(9 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ $\delta 199.4,174.3,144.8,139.5,133.7,122.8,60.0,47.3,41.8,34.8,27.8$, $27.6,26.1,25.5,21.8$ and 14.1 ; IR $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$, solution) $v / \mathrm{cm}^{-1} 3072 \mathrm{w}$, $2995 \mathrm{~m}, 1723 \mathrm{~s}, 1659 \mathrm{~s}, 1601 \mathrm{~m}, 1427 \mathrm{~m}, 1375 \mathrm{~m}$ and $1182 \mathrm{~s} \mathrm{~cm}^{-1}$; mass (EI) $\mathrm{m} / \mathrm{z} 262\left(\mathrm{M}^{+}, 77 \%\right), 216(23), 188(37), 145(100), 105(9), 91$
(17); high resolution, calc. for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{3} 262.1563$; found $\mathrm{m} / \mathrm{z}$ 262.1567.

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