Construction of Fused Bicyclo-[5.3.0]decane and -[5.4.0]undecane Ring Skeletons via Sequential Additions of Nucleophiles to Tricarbonyltropyliumchromium Tetrafluroborate

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The addition of a variety of the highly functionalized zinc–copper reagents RCu(CN)ZnI to the title cation gives $(\eta^{6}$ -cyclohepta-1,3,5-triene)Cr(CO)₃ complexes with a functionalized side chain at the C-7 position of the ring: intramolecular cyclisation of ester-substituted adducts using lithium diisopropylamide generates fused bicyclo-[5.3.0]decane and -[5.4.0]undecane derivatives.

Seven-membered carbocycles are often used in the construction of complicated natural products.1 Specially, the perhydroazulene or bicyclo[5.3.0]decane ring system is the most commonly encountered of the cycloheptane derived structure occurring biologically-active compounds.2 natural in Recently, tropone has been shown to be a versatile building block for the synthesis of a variety of sesquiterpenes with the bicyclo[5.3.0]decane ring skeleton.³ Normally, several steps starting from the 1,8-addition of a protected functionalized three-carbon chain to tropone are needed to construct the bicyclo[5.3.0]decane ring skeleton. However, bicyclo-[5.3.0]decane and -[5.4.0]undecane derivatives may be available easily via sequential additions of nucleophiles to [(tropyl $ium)Cr(CO)_3$ cation 1. This method allows the formation of 5,7- and 6,7-cis-fused bicyclic systems which cannot be generated by sequential additions of nucleophiles to



 $(\eta^{5}$ -pentadienyl)Fe(CO)₃.⁴ Prior to our investigation, several classes of nucleophiles such as hydrides, methoxides and hydrogen sulfides had been shown to add to cation 1 to produce 7-exo-substituted (n⁶-cyclohepta-1,3,5-triene)-Cr(CO)₃ complexes.⁵ However, the addition of an excess of sodium diethylmalonate or lithium cyclopentadienide to 1 provided the ring contraction product [C₆H₆Cr(CO)₃]⁶ and the reaction of cyanide with 1 led by reductive coupling to a mixture of bi(cycloheptadienyl)Cr₂(CO)₆ and bi(cycloheptadienyl)Cr(CO)₃.⁷ Moreover, phenyl lithium failed to react with 1. Surprisingly, reports on the addition of organocopper derivatives to cation 1 are lacking. We now report that the highly functionalized zinc-copper reagents [RCu(CN)ZnI]^{4,8} add efficiently to 1 to afford (η^6 -cyclohepta-1,3,5-triene)- $Cr(CO)_3$ complexes 2a-h with a functionalized side chain at the C-7 position of the ring. The reactions are performed in the expectation that further manipulation of the resulting complexes will then be possible after the initial additions.

Cation 1 was prepared in two steps from cyclohepta-1,3,5triene and Cr(CO)₃(MeCN)₃ (85% overall yield).⁹ Finally, our synthesis of C-7-substituted (n⁶-cyclohepta-1,3,5-triene)- $Cr(CO)_3$ complexes 2 involved the slow addition of a THF solution of the highly functionalized zinc-copper reagents RCu(CN)ZnI (1.2 equiv.) to a stirred suspension of cation 1 in THF at 0 °C under nitrogen. The addition was carried out for 5 h at 23 °C followed by workup with saturated aqueous ammonium chloride and diethyl ether extraction. After purification by flash column chromatography on silica gel, complexes 2a-h were obtained as the major products and in good yields (69-96%, Scheme 1). The trans relationship between the nucleophile and the metal moiety of complexes 2 was assigned based upon comparison of their ¹H NMR spectra $[\delta 5.95 (2H, br.s)]$ with the spectra of known 7-exo-(η^{6} cyclohepta-1,3,5-triene)Cr(CO)₃ complexes.¹⁰ Further manipulation of the resulting complexes 2 was demonstrated as follows. Treatment of 2b with lithium diisopropylamide (LDA, 1.2 equiv.) in THF and hexamethylphosphoramide (HMPA, THF-HMPA = 3:1) at -78 °C for 1 h followed by quenching the reaction mixture with TFA (10 equiv.) produced a mixture of perhydroazulene derivatives 3a and 3b in a 1:1 ratio in 77% total yield. The formation of bicyclo[5.3.0]decane derivatives 3a and 3b agrees closely with the mechanism proposed for the intermolecular addition of nucleophiles to $(\eta^{6}-arene)Cr(CO)_{3}$.¹¹ Deprotonation of **2b** using LDA at -78°C in THF-HMPA would result in the formation of the kinetic ester enolate.4,12 Anti addition of the ester enolate to the terminal position of the triene ligands gave $(\eta^{5}$ -cycloheptadienyl)Cr(CO)₃ anion complex 4 (n = 1, Fig. 1). Electrophilic quenching of 4 with TFA afforded bicyclo[5.3.0]decane derivatives 5a and 5b. However, diene migration occurred and produced thermally more stable isomers 3a and 3b under the reaction conditions. Treatment of 2c[†] with LDA (1.2 equiv.) in THF-HMPA (3:1) under at atmosphere of CO at -78 °C for 1 h followed by quenching the reaction mixture with iodomethane (5 equiv.; CO; 25 °C; 2 h), and then iodine (4 equiv.; 25°C; 4 h) generated the bicyclo[5.4.0]undecane derivative 6† as the only diastereoisomer isolated in 63% yield. Reaction of 4 (n = 2) with iodomethane under an atmosphere of CO would give the bicyclo[5.4.0]undecane derivative 713 which isomerized to provide the highly conjugate dienone 6. The cis stereochemistry at the ring juncture of 6 is fixed by anti addition of the ester enolate to the terminal position of the triene ligand (to give 4, n = 2) and the ester group assigned as *exo* was resulted from the formation of the kinetic ester enolate under the kinectically controlled reaction conditions.⁴ ¹H NMR studies provided the initial evidence for support of the structural assignments. The ¹H NMR spectrum of compound 6 exhibited a doublet of doublets, centred at δ 3.2, J = 12.0, 2.8Hz, assigned to H_a. The coupling constant of H_a-H_b (J_{ab}) of 12.0 Hz agrees closely with the 12.7 Hz coupling constant observed for the cis fused bicyclo[5.4.0]undecanone and the coupling constant of H_a-H_c (J_{ac}) of 2.8 Hz agrees with the 3.2 Hz coupling constant observed for the similarly disposed *trans* hydrogens.14

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† Spectroscopic data for: 2c: ¹H NMR (CDCl₃) δ 5.95 (2H, brs), 4.77 (2H, brs), 4.10 (2H, q, J 7.2 Hz), 3.70 (2H, t, J 8.6 Hz), 2.90 (1H, m), 2.17 (2H, t, J 7.3 Hz), 1.39 (2H, m), 1.23 (3H, t, J 7.2 Hz), 1.06 (2H, m), 1.24 (2 J, 2 Hz), 1.06 (2H, m), 1.25 (2H, m), 1.2 m) and 0.25 (2H, m); ¹³C NMR (CDCl₃) & 231.9, 173.4, 98.9, 97.8, 67.7, 60.2, 38.6, 35.9, 34.0, 24.3, 23.2 and 14.2; IR (CH₂Cl₂, solution) v/cm⁻¹ 3067w, 2992w, 1977s, 1910s, 1879s, 1728w, 1620, 1424m and 1252m; mass (EI) m/z 357 (M+, 27%), 328 (8), 300 (5), 272 (73), 254 (11), 194 (46), 180 (100); high resolution, calc. for $C_{17}H_{20}O_5Cr$ 356.0710; found *m/z* 356.0712. For **6**: ¹H NMR (CDCl₃) δ 6.84 (1H, d, J 7.8 Hz), 6.14 (1H, dd, J 10.7, 8.3 Hz), 5.91 (1H, m), 4.14 (2H, J 6.8 Hz), 3.26 (1H, dd, J 12.0, 2.8 Hz), 2.71-2.61 (2H, m), 2.44 (1H, m), 2.34 (3H, s), 2.12 (1H, m) and 1.82-1.23 (9H, m); ¹³C NMR (CDCl₃) δ 199.4, 174.3, 144.8, 139.5, 133.7, 122.8, 60.0, 47.3, 41.8, 34.8, 27.8, 27.6, 26.1, 25.5, 21.8 and 14.1; IR (CH₂Cl₂, solution) ν /cm⁻¹ 3072w, 2995m, 1723s, 1659s, 1601m, 1427m, 1375m and 1182s cm⁻¹; mass (EI) m/z 262 (M⁺, 77%), 216 (23), 188 (37), 145 (100), 105 (9), 91 (17); high resolution, calc. for $C_{16}H_{22}O_3$ 262.1563; found m/z262.1567.

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